Attributes of the Homestake Mine for Investigation of Reaction-Transport Processes



- Validation of coupled reaction-transport conceptual and numerical models requires well-controlled in-situ experiments that could be sited at depth at Homestake
- The Fe- and Mg-rich mineral assemblage (e.g., cummingtonite, chlorite, Fe-Mg carbonates, graphite) is strikingly different from that in the rhyolitic tuffs and granites that are the host rocks for other thermal tests
- Effective reaction rates are controlled by the hierarchy of scale of fluid flow, geologic structure, and mineral fabric, all of which are very different in the highly deformed metamorphic rocks at the Homestake Mine compared to the welded tuffs at YM or that in plutonic rocks
- Reactions of ore minerals such as pyrrhotite, arsenopyrite, and gold with hydrothermal waters will allow for validation of numerical models of coupled water-rock interaction and metal transport

Geochemical Attributes

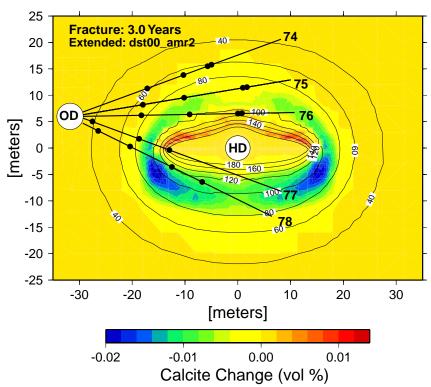


- Owing to the age of the rocks (~ 2.2 billion years old), the ⁸⁷Sr/⁸⁶Sr ratios of the minerals will be very different owing to the different initial Rb/Sr ratios. Therefore, shifts in the ⁸⁷Sr/⁸⁶Sr ratio in waters could be used to constrain mineral reaction rates
- For example, reaction of hydrothermal waters with calcite would be rapid, compared to silicate minerals, but would likely have a lower ⁸⁷Sr/⁸⁶Sr ratio owing to a potentially low initial Rb content
- Reaction-transport models of thermal tests at YM have shown areas of calcite dissolution strongest in areas of fracture drainage below the heaters (see following slide), which is supported by Sr contents of waters and C-14 analyses of gases. However, there is little difference in ⁸⁷Sr/⁸⁶Sr ratios between the waters and the calcite because of the origin of the calcite from percolating meteoric waters, and the relatively young ages of the rocks

Reaction-Transport Models and Data from YM Thermal Tests







Gas Compositions

